1. Identification of the Substance/Mixture and of the Company/Undertaking:

1.1 **Product Identifier:** Lithium Magnesium Alloy

1.1.1 **Substances**

Not applicable

1.1.2 **Mixture name:** Lithium Magnesium Alloy

**Alternate names and trade name**

Lectro® Max 130 Anode Material, Lithium Magnesium Alloy, ingot, foil, parts

1.2 **Relevant Identified Uses of the Substance or Mixture and Uses Advised Against:**

Formulation and chemical synthesis in industrial manufacturing operations;

Additive for preparations and articles for industrial and consumer use;

Component in lithium batteries.

Do not use for private purposes (household).

1.3 **Details of the Supplier of the Safety Data Sheet**

**North America**

FMC Lithium

Seven LakePointe Plaza

2801 Yorkmont Rd, Suite 300

Charlotte, NC 28208

Phone: +1.704.868.5300

Fax: +1.704.868.5370

1.888.lithium

Email: lithium.info@fmc.com

Web: www.fmclithium.com

**Europe**

FMC Chemicals

Commercial Road

Bromborough, Merseyside

CH62 3NL, England

Phone: +44.151.334.8085

Fax: +44.151.482.7361

**Asia Pacific**

FMC Asia Innovation Center

No 3 Building No. 4560

Jinke Road

Shanghai, China 201203

T: +86.21.2067.5888

1.4 **Emergency Telephone Number:**

**North America**

CHEMTREC: +1.800.424.9300

+1.703.527.3887

Plant: +1.704.629.5361

Medical: +1.303.595.9048

**Europe**

24 hr Specialist advice number:

CHEMTREC: +1.703.527.3887

Office (0900-1700): +44.151.334.8085

**Asia Pacific**

Phone: +86.21.2067.5888

2. Hazards Identification

2.1 **Classification of the Substance or mixture:**

**2.1.1 GHS Classification** [EC Regulation No 1272/2008 and US OSHA regulations]

- Water-reactive, Category 1
- Corrosive to skin, Category 1B
- Eye damage; Category 1

**2.2.2 EC: Classification according to 67/548/EEC or 1999/45/EC [DSD/DPD]**

- F, R14/15; C, R34

2.2 **Label Elements:**

**2.2.3 Hazard Pictograms(s):**

[Diagram]

**2.2.4 Signal Word:** Danger

**Hazard Statement(s):**

- In contact with water releases flammable gases which may ignite spontaneously
- Causes severe skin burns and eye damage

**Precautionary Statement(s):**

- Keep away from any possible contact with water, because of violent reaction and possible flash fire.
Handle under inert gas. Protect from moisture.

Wear protective gloves/protective clothing/eye protection/face protection.

In case of fire: Use graphite, copper powder, Lith-X (Ansul) for extinction.

Rinse skin with water/shower.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

Immediately call a POISON CENTER or doctor/physician.

Additional Precautionary Statement(s):

Do not breathe dust/fume/gas/mist/vapours/spray.

Wash hands thoroughly after handling.

IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.

Wash contaminated clothing before reuse.

Store in a dry place. Store in a closed container.

Dispose of contents/ container to an approved waste disposal plant.

2.3 Other Hazards

Reacts violently with water EUH014

3. Composition / Information on Ingredients

3.1 Substances

Not applicable.

3.2 Mixtures

3.1.1 GHS Classification [EC: Regulation No 1272/2008; US: OSHA regulations]

<table>
<thead>
<tr>
<th>Chemical Name</th>
<th>CAS #</th>
<th>EC No</th>
<th>EC Index No</th>
<th>REACH Reg No</th>
<th>Wt.%</th>
<th>Classification, Hazard Statement Codes</th>
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<td>231-102-5</td>
<td>003-001-00-4</td>
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<td>F R15 H260 H314 EUH014</td>
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<td>Not available</td>
<td>5-30</td>
<td>F R15 H251 H260</td>
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3.1.2 EC: Classification according to 67/548/EEC or 1999/45/EC [DSD/DPD]

<table>
<thead>
<tr>
<th>Chemical Name</th>
<th>CAS #</th>
<th>EC No</th>
<th>Wt.%</th>
<th>Symbols</th>
<th>R-phrases</th>
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<td>70-95</td>
<td>F</td>
<td>R15 R14 R34</td>
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<td>magnesium</td>
<td>7439-95-4</td>
<td>231-104-6</td>
<td>5-30</td>
<td>F</td>
<td>R15 R14</td>
</tr>
</tbody>
</table>

(see Section 16 for R-phrase text)

4. First Aid Measures

4.1 Description of First Aid Measures

EYES: Immediately flush with water for at least 15 minutes, lifting the upper and lower eyelids intermittently. See a medical doctor or ophthalmologist immediately.

SKIN: Quickly wipe off as much as possible, then immediately flush with plenty of water while removing contaminated clothing and/or shoes. Thoroughly wash with soap and water. Obtain immediate medical attention. Contact a medical doctor if necessary.

INGESTION: Quickly wipe material from the mouth and rinse mouth with water. Do not induce vomiting.

Never give anything by mouth to an unconscious person. See a medical doctor immediately.

INHALATION: Remove to fresh air. If breathing difficulty or discomfort occurs and persists, see a medical doctor. If breathing has stopped, give artificial respiration and see a medical doctor immediately.

4.2 Most Important Symptoms and effects, both acute and delayed

Lithium magnesium alloy reacts violently with water and is corrosive to the eyes, skin and respiratory tract. Treatment should first remove as much of the material as possible as
4.3 Indication of any immediate medical attention and special treatment needed.

Notes to medical doctor:
This product is corrosive and reacts violently with water. Treatment should first remove as much of the material as possible as quickly as possible, then flush with very large quantities of water. Ingestion presents a singular problem as emesis may produce esophageal damage and/or aspiration damage; dilution with water or other water-containing materials may produce a reaction that exacerbates the corrosive activity. Consideration may be given to gastric lavage with a large diameter tube for removal of material and then dilution with large amounts of water. Esophagoscopy may be of assistance in this procedure and to assess extent of damage. Treatment is otherwise symptomatic and supportive.

5. Fire-Fighting Measures

5.1 Extinguishing media
DO NOT USE WATER, SAND OR CARBON DIOXIDE. Use graphite, copper powder, Lith-X (Ansul). If not available, dry sodium chloride, dry (anhydrous) calcium oxide or dry lithium chloride can be used.

5.2 Special hazards arising from the substance or mixture

Hazardous combustion products
Lithium and magnesium oxide, lithium and magnesium hydroxides
General Hazard
Flammable solid. Water reactive.
Properties contributing to Flammability
Water reactivity of solid and flammable hydrogen gas given off on reaction with moisture.
Flashpoint
Not applicable
Flammable limits in air
Not applicable
Auto ignition temperature
At temperatures above the melting point (180.5°C) lithium metal can catch fire spontaneously on contact with air.

Sensitivity to static discharge
Not applicable
Sensitivity to static impact
Not applicable

5.3 Advice for fire-fighters
Wear full protective clothing and self-contained breathing apparatus (SCBA) approved for fire fighting. This is necessary to protect against the hazards of heat, products of combustion and oxygen deficiency. Do not breathe smoke, gases or vapors generated.

Lithium fires can throw off molten lithium metal particles. Burning lithium releases corrosive lithium oxide dust and fumes. Lithium metal can reignite after fire is initially extinguished. Never leave extinguished fire unattended. After all material has apparently burned and cooled, carefully turn over remaining residue and be prepared to reextinguish should reaction occur. Carefully place residue in steel drum, using a long-handled shovel, and cover with extinguishing media.

For additional fire fighting information, see National Fire Protection Assn. Standard NFPA 485.

COMMENTS:
(See Section 10, Stability and Reactivity)

6. Accidental Release Measures

6.1 Personal precautions, protective equipment and emergency procedures
Before cleanup measures begin, review the entire SDS with particular attention to Section 2, Hazards Identification; and Section 8, Exposure Controls/Personal Protection.

6.2 Environmental precautions
Do not wash into drains. Dispose of at a qualified waste disposal facility.

6.3 Methods and material for containment and cleaning up
Remove all sources of ignition. To prevent ignition, cover with mineral oil (or kerosene), soaking thoroughly, and place in oiled steel drums which are approved for transport. Keep water and moisture away from spilled material. Dispose of waste according to local and Federal laws and regulations.

6.4 Reference to other sections
Before cleanup measures begin, review the entire SDS with particular attention to Section 2, Hazards Identification; and Section 8, Exposure Controls/Personal Protection.

6.5 Additional information
Not specified.

7. Handling and Storage
7.1 **Precautions for safe handling**

Wear safety glasses or goggles and dry rubber gloves. Where relative humidity is maintained below 50%, or lithium magnesium alloy surface is coated with mineral oil, pieces can be handled in open atmosphere at room temperature. To maintain best quality, humidity levels of less than 2% are recommended.

7.2 **Conditions for safe storage, including any incompatibilities**

Store in original unopened shipping container. Once opened, store in argon atmosphere or mineral oil. Keep away from water, humid air, acids and oxidizing materials. Keep away from heat, sparks and flame.

7.3 **Specific end use(s)**

Defined in Exposure scenarios. Industrial and professional use only

8. Exposure Controls / Personal Protection

8.1 **Control parameters**

**Lithium metal**

**DNEL**

- Long-term exposure, systemic, inhalation: 4.2 mg/m³
- Long-term exposure, systemic, dermal: 12 mg/kg/day

**PNEC**

- PNEC aqua (freshwater): 0.165 mg/l
- PNEC aqua (freshwater, intermittent): 1.65 mg/l
- PNEC STP: 23 mg/l

**EXPOSURE LIMITS**

<table>
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<th>Chemical Name</th>
<th>EU (TWA)</th>
<th>EU (STEL)</th>
<th>EH40 (UK WEL) TWA</th>
<th>EH40 (UK WEL) STEL</th>
<th>USA (ACGIH) TWA</th>
<th>USA (ACGIH) STEL/Ceiling</th>
<th>USA (OSHA) PEL</th>
<th>USA (OSHA) STEL/Ceiling</th>
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</thead>
<tbody>
<tr>
<td>lithium</td>
<td>none*</td>
<td>none*</td>
<td>none*</td>
<td>none*</td>
<td>none*</td>
<td>none*</td>
<td>none*</td>
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</tr>
</tbody>
</table>

* No occupational exposure limit value

8.2 **Exposure controls**

**Engineering controls:** Use local exhaust ventilation to keep airborne concentrations below exposure limits.

**Personal protective equipment**

**Eyes and Face:** Safety glasses or goggles for general use. Full flame-resistant face shield required if metal is in a molten state.

**Respiratory:** None

**Protective Clothing:**

- **Eyes:** Safety glasses or goggles for general use. Full flame-resistant face shield required if metal is in a molten state.
- **Gloves:** Dry rubber gloves for general use. Wear full flame-resistant clothing if the metal is handled or used in a molten state.
- **Other:** Quick-drench eyewash and safety shower.

**Work Hygienic Practices:**

Quick-drench eyewash and safety shower.

9. Physical and Chemical Properties

9.1 **Information on basic physical and chemical properties**

- **Appearance:** Silvery-white soft metal solid
- **Odor:** Odorless
- **Odor threshold:** Not applicable
- **pH:** Reaction with water creates a highly alkaline solution pH>12
- **Melting point:** For Li metal: 180.5°C
- **Boiling point:** For Lithium metal: 1317°C
- **Flash point:** Not applicable
- **Evaporation rate (butyl acetate = 1):** Not applicable
- **Flammability:** Reacts with water to produce hydrogen, a flammable gas.
- **Flammable limits:** Not applicable
- **Vapor pressure:** Not applicable
- **Vapor density (air = 1):** Not applicable
Specific gravity: 0.61 g/cc, 0.5 g/cc (Li)
Solubility in water: Reacts violently with water
Partition coefficient n-octanol/ water: Not applicable
Autoignition temperature: At temperatures above the melting point (180.5°C) lithium metal can catch fire spontaneously on contact with air.
Decomposition temperature: Not available
Viscosity: Not available
Explosive properties: Not explosive
Oxidizing properties: Not an oxidizer

9.2 Other information
   Self-reactive properties
   Pyrophoric properties
   Under normal working conditions, does not meet classification criteria.
   Self-heating properties
   Water reactive properties
   Corrosive to metals
   Molecular weight: 6.94 (Li) 24.305 (Mg)

10. Stability and Reactivity
10.1 Reactivity
Reacts violently with water, producing flammable hydrogen gas.
10.2 Chemical stability
Stable when kept dry and under inert gas.
10.3 Possibility of hazardous reaction
Reacts violently with water, producing flammable hydrogen gas. Hazardous polymerization will not occur.
10.4 Conditions to avoid
Temperatures above the melting point (for Li, 180.5°C/357ºF), and contact with water, moisture or humid air.
10.5 Incompatible materials
Acids, oxidizers, oxygen, nitrogen, or carbon dioxide
10.6 Hazardous decomposition products
Lithium is an element and does not decompose. However, it is highly reactive in contact with many other substances, releasing large quantities of heat and/or hazardous products. It can react violently with water, the humidity in air, and the moisture in other substances, releasing hydrogen gas, which may catch fire explosively. Corrosive fumes of lithium oxide and/or lithium hydroxide are also released.

11. Toxicological Information
11.1 Information on toxicological effects
   (a) acute toxicity
   Based on the available data, the classification criteria are not met.
   (b) skin corrosion/irritation
   Classified as corrosive to skin on the basis of lithium.
   (c) serious eye damage/irritation
   Classified as corrosive to eyes on the basis of lithium.
   (d) respiratory/skin sensitisation
   Classed as not sensitizing to skin on the basis of lithium.
   (e) germ cell mutagenicity
   Classified as not mutagenic based on lithium.
   (f) carcinogenicity
   Classified as not carcinogenic based on lithium.
   (g) reproductive toxicity
   Classified as not a reproductive toxin based on lithium.
   (h) STOT-single exposure
   Classified as not causing organ damage based on lithium.
   (i) STOT-repeated exposure
   Classified as not causing organ damage on repeat exposure based on lithium.
   (j) aspiration hazard
   Lithium magnesium alloy, a solid, does not present an aspiration hazard.

Lithium has been extensively tested for REACH registration

Acute Effects From Overexposure:
This product is extremely reactive with body moisture and is corrosive to skin, nose, throat, stomach and eyes (may cause blindness).

Chronic Effects From Overexposure:
No data available for product.
Carcinogenicity Listings

EH40: Not listed.
IARC: Not listed.
NTP: Not listed.
OSHA: Not considered a carcinogen under OSHA.
ACGIH: Not listed.

12. Ecological Information

12.1 Toxicity: No classification.
Lithium
- Fish, short-term, freshwater: LC50 = 18 mg/L
- Fish, long-term, freshwater: NOEC = 2.87 mg/L
- Daphnia magna, short-term, freshwater: EC50 = 10 mg/L with pH-adjustment
- Daphnia magna, long-term, freshwater: NOEC = 1.7 mg/L
- Algae (Pseudokirchneriella subcapitata), long-term, freshwater: ErC50 = 25.6 mg/L

12.2 Persistence and degradability
Not applicable for metal.
Material reacts slowly with air in the environment to form lithium and magnesium hydroxides, lithium and magnesium carbonates and nitrides.

12.3 Bioaccumulative potential
Not accumulative

12.4 Mobility in soil
No data available for the product.

12.5 Results of PBT and vPvB assessment
Not applicable for metal.

12.6 Other adverse effects
Lithium magnesium alloy reacts violently with water. The hydrolysis products consist of hydrogen gas and lithium hydroxide. The hydroxide ion may affect the pH of the water.

13. Disposal Considerations

13.1 Waste treatment methods
Waste containing lithium magnesium alloy is considered a reactive waste. Disposal facilities specializing in the handling of reactive waste are recommended. Dispose of waste according to local and Federal laws and regulations.

14. Transport Information

14.1 UN Number
UN3208

14.2 UN proper shipping name (IMDG, ICAO, ADR, DOT)
Metallic substance, water reactive, N.O.S (lithium magnesium alloy)

14.3 Transport hazard class(es) (IMDG, ICAO, ADR, DOT)
4.3, Dangerous when wet

14.4 Packing group (IMDG, ICAO, ADR, DOT)
I

14.5 Environmental hazards
Based on available data, the classification criteria are not met.

14.6 Special precautions for user
None

14.7 Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code
Based on available data, the classification criteria are not met.

15. Regulatory Information

15.1 Safety, health and environmental regulations/legislation specific for the substance [or mixture]

EUROPEAN UNION:

German Wassergefährdungsklasse (water hazard class)
lithium 2
magnesium not available

UNITED STATES:
Section 311 Hazard Category (40 CFR 370): Reactive, fire hazard, immediate (acute) health hazard.

Section 313 Reportable Ingredients (40 CFR 372):
This product does not contain a toxic chemical subject to the reporting requirements of Section 313 of Emergency Planning and Community Right-To-Know Act of 1986.

Section 302 Extremely Hazardous Substances (40 CFR 355): Not listed

CERCLA Hazardous Substance (40 CFR 302.4): Not listed

TSCA Sec 12b Export Notification: This product is not subject to TSCA 12 (b) Export Notification Requirements.

NFPA Rating:

INTERNATIONAL INVENTORY STATUS:

<table>
<thead>
<tr>
<th>Inventory/Country</th>
<th>Product Status</th>
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<td>TSCA (US)</td>
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<tr>
<td>DSL (Canada)</td>
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</tbody>
</table>

15.2 Chemical Safety Assessment
A Chemical Safety Assessment has been completed for lithium metal.

16. Other Information

European Union:

R Phrases:
R14/15 Reacts violently with water, liberating extremely flammable gases
R34 Causes burns

List of Abbreviations used in this SDS:

PBT Persistent, Bioaccumulative and Toxic
vPvB very Persistent, very Bioaccumulative
PEC Predicted environmental concentration
PNEC Predicted no effect concentration
DNEL Derived no effect level

Specific uses identified for Exposure Scenarios for lithium metal:

ES1 Industrial use organolithium production and pharma synthesis
ES2 Industrial use, battery foils
ES3 Industrial use Li/Al alloys
ES4 Professional laboratory use

REVISION SUMMARY:
Revision # 0. New SDS.

This SDS has been prepared to meet U. S. OSHA Hazard Communication Standard requirements.

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